

Silver Diamine Fluoride (SDF): The Miracle Fluid

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Abstract

Although there has been a decrease in the prevalence and the severity of dental caries in children over the past few decades, still, various studies in literature worldwide report high prevalence of dental caries. Traditional restorative methods for treating dental caries are not always affordable or possible, as it requires patient cooperation for desirable outcome. To overcome this problem, Silver diamine fluoride (SDF) was first promoted in Japan in late 1960's. The antimicrobial effect of silver compounds has been proven by the 100-year-old application of silver compounds for the prevention and treatment of various infections. Now in many countries a 38% (44,800 ppm fluoride) SDF solution is commonly used to arrest caries and to reduce hypersensitivity. Application of SDF to arrest dental caries is a non-invasive procedure that is quick and simple to use. When SDF is applied on carious lesions, the fluoride enhances remineralization and the silver ions act as an antibacterial agent and inhibit the growth of cariogenic biofilms. However, it has certain drawbacks like black discoloration of the carious teeth and an unpleasant metallic taste. But, the low cost of SDF and its simplicity in application suggest this as a novel preventive agent in public dental health projects.

Keywords: Silver diamine fluoride; Silver; Dental caries; Preventive dentistry.

Introduction

Dental caries is the quotidian pandemic, inflicting all ages of the world's population.¹ Dental caries is a bacterial driven, generally chronic, site-specific, multifactorial, dynamic disease process that results from the imbalance in the physiologic equilibrium between the tooth mineral and the plaque fluid; that is, when the pH drop results in net mineral loss over time.²

However, research has shown that caries can be prevented or arrested by timely approach and action. Multitude of evidence-based approaches for caries prevention have been reported; however, these strategies demand significant financial investment and depend on the availability of oral health workforces and facilities.³

Preventive measures include the use of fluoride varnish—such as 5% sodium fluoride (NaF)—and the use of fluoridated toothpaste⁴⁻⁷ and atraumatic

restorative treatment (ART). ART is painless, low cost, and can be applied outside the clinical setting or when conventional treatment is not available. High rate of failure is one of the major drawbacks of Silver topical products, such as silver nitrate.⁸

SDF have been used in Japan for over 40 years to arrest caries and reduce tooth hypersensitivity in primary and permanent teeth. During the past decade, many other countries such as Australia and China have started using this compound with similar success.^{9,10}

The exact mechanism of SDF is not well demarcated. It is predicted that fluoride ions act mainly on the tooth structure, while silver ions, like other heavy metals, exert antimicrobial action. Further, SDF reacts with hydroxyapatite in an alkaline environment to form calcium fluoride (CaF₂) and silver phosphate as major reaction products. CaF₂ provides a fluoride reservoir to form fluorapatite which is less soluble than hydroxyapatite in an acidic environment.^{11,12}

Nevertheless, the black staining of the lesions associated with the application of SDF is inevitable. In the United States, SDF is marketed as 38 percent silver diamine fluoride which is equivalent to five percent fluoride in a colorless liquid, with a pH of 10. SDF reportedly also has approximately 2-3 times more fluoride retained than delivered by sodium fluoride, stannous fluoride, or acidulated phosphate fluoride (APF) commonly found in foams, gels, and varnishes.¹³ Also, SDF has not been shown any deterrent effect on adhesion of resinor glass ionomer restorative material^{11,14,15}

SDF has shown promising results in a horde of studies. Thus, this systematic review is an attempt to provide an insight to mechanism and effectiveness of SDF in preventive and therapeutic dentistry.

Method of literature search

A literature search of PubMed®/MEDLINE and Google Scholar databases were conducted using the terms: diamine silver fluoride and caries, and silver diamine fluoride. A total of around 83 articles were found. Inclusion criteria were all *in vivo* and *in vitro* studies in which SDF was used as a study group as well as review articles on SDF. Exclusion criteria were: early reports of longer studies, editorials, (with exceptions of few articles in Chinese and Japanese journals that included early study and history of SDF).

Milestones of SDF

Silver (Ag) has been documented to be used for medicinal purposes dating back to 1000 bc for storage of water. ¹⁶ Before silver diamine fluoride (SDF), silver nitrate followed by fluoride varnish was the only noninvasive option available for caries treatment. ¹⁷ Case series of carious lesions arrested by silver nitrate date to the 19 th century, where 87 of 142 treated lesions were arrested. ¹⁸
Around AD 1000, in Japan, there was a custom among the ladies to dye their teeth black called "Ohaguro," for expressing married. Although, it was tooth cosmetics, ought to prevent dental caries. ¹⁹
SDF has been accepted as a therapeutic agent by the Central Pharmaceutical Council of the Ministry of Health and Welfare in Japan for dental treatment since the 1960s. ²⁰
Craig <i>et al.</i> reported that AgF solution was used in dentistry as early as the 1970s. ²¹
In 1891, silver amalgam and nitric acid was used on carious teeth and had caries arresting properties ¹⁸ Silver nitrate was directly applied to carious cavities with analogous outcomes, and it was termed as Howe's solution, which was used for caries inhibition. ²²
A solution of 38% SDF was also used in China to arrest caries. ²³
Also, a few dentists in Southern California who used SDF to arrest caries of young children with early childhood caries. ²⁴
SDF has been widely used in Australia and Brazil since 1980 with <i>in vivo</i> studies done in the respective countries. ^{25,26} Interest in SDF has resurfaced in recent times with several studies proving its effectiveness in preventing and arresting caries in the primary dentition and first permanent molar in children. ²⁷⁻³²
The need for anticaries agents such as SDF is perhaps best understood in terms of the World Health Organization (WHO) Millennium Development Goals for Health and particularly oral health. ³³
In Western Australia, 40% silver fluoride (AgF) was applied as the treatment for deep dental caries in primary teeth at school dental care services. ¹⁹
For past few decades, SDF has also been used in Australia and China to prevent dental decay. ^{34,35}
SDF has been used in many community dental health programs in various concentrations has been used in Argentina, Brazil and Spain; and additionally community health programs were designed for sub-Saharan Africa and for other countries of Africa. ³⁶
In 2005, SDF was approved by the US Food and Drug Administration for management of tooth hypersensitivity. ³⁷
Its extensive use started in China in start of 21 st century as a caries arresting agent in school going children. ^{13,29}
From 2005 to 2009 in Australia, a series of <i>in vitro</i> studies were conducted and proved effect of SDF on <i>Streptococcus mutans</i> and dental bio-film as acaries arresting and antimicrobial agent respectively. ³⁸⁻⁴¹
In 2013, SDF was used to prevent root caries in elderly. ⁴²
Studies were conducted in India providing literature of successful use of SDF as a caries arresting agent. ^{43,44}
A study in 2018 found that SDF was successful in arresting active caries in primary teeth of young children in USA and was also well received by their parents. ⁴⁵

Mechanism of action of silver

Multiple modes of action have been proposed for silver. Studies have indicated that silver interacts with sulfhydryl groups of proteins and with deoxyribonucleic acid (DNA), altering hydrogen bonding and inhibiting respiratory processes, DNA unwinding, cell-wall synthesis, and cell division.^{23,46} At the macro level these interactions effect bacterial killing and inhibit biofilm formation.⁴⁷

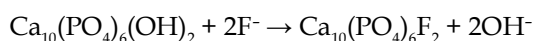
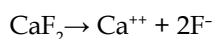
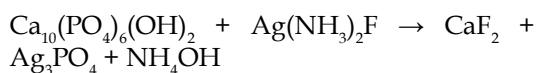
Three possible mechanisms of action of SDF on caries as described by Shimizu and Kawagoe are:⁴⁹

1. Obturation of dentinal tubules.

It has been proven that invasion of dentinal tubules by microorganisms is the hallway for caries, thus obturation of these organic roads can lead to prevention of dental caries⁵⁰ Shimizu evinced that dentin treated with SDF showed increased electric resistance and decreased dye permeability⁵² because silver and its compounds blocked the peritubular and intertubular dentin, oligodynamic action of silver lead to inhibition of microorganisms^{52,53} demonstrated that the use of 38% SDF inhibited demineralization and preserved collagen from degradation in demineralized dentin.

2. Cariostatic action of the reaction products between SDF and mineral component of the tooth.

F⁻ ion of SDF applied to dentin under *in vivo* conditions penetrated to a depth of 50–100 μ .^{54,55} It has been reported that SDF (Ag(NH₃)₂F) reacts with the tooth mineral hydroxyapatite (HA)(Ca₁₀(PO₄)₆(OH)₂) to release calcium fluoride (CaF₂) and silver phosphate (Ag₃PO₄), which are responsible for the prevention and hardening of dental caries. A simplified chemical reaction was suggested as shown below^{12,13}



3. Antienzymatic actions of the reaction products between Ag(NH₃)₂F and organic component of the tooth.

Its antibacterial properties arise from inhibition of the enzyme activities and dextran-induced agglutination of cariogenic

strains of *Streptococcus mutans*.⁵⁶ Sunada *et al.*⁵⁷ found that dentin, which had been treated with Ag(NH₃)₂OH by ionophoresis increased in resistance to trypsin and stated that it-might be owing to reaction of Ag and organic component of dentin. Yanagida *et al.*⁵⁸ showed that dentin protein treated with Ag(NH₃)₂F had increased in resistance to collagenase and trypsin. These actions of Ag and Ag(NH₃)₂F to organic component of the tooth may also be contribute to inhibit caries.

Suzuki *et al.*⁵⁶ studied the mechanism of antiplaque action of diamine silver fluoride [Ag(NH₃)₂F] and indicated that silver ion may inhibit the colonization of *S. mutans* on enamel surface and offer a possible explanation for the antiplaque action of the agent. It was also shown that the binding of glucan to HA was inhibited by the treatment of HA with fluoride solution, but was slightly promoted by that with silver solution. The adsorption of salivary proteins by HA was inhibited by the treatment of HA with Ag(NH₃)₂F. This was due to both fluoride and silver ions. Thus, the potential to inhibit the *S. mutans* is more for SDF than any other fluoridated solution.

Also, Mei *et al.* (2013) mentioned that 38% SDF inhibits multi-species cariogenic biofilm formation on dentin carious lesions and reduces the demineralization process.⁵⁴ Suzuki *et al.* showed that after SDF application, fluoride ions penetrated the enamel diffusely to about 25 μ , whereas silver ions were mainly deposited on the surface and some penetrated as far as 20 μ into the enamel.⁵⁹

Clinical applications of Silver Diamine Fluoride

Prevention and arrest of early childhood caries

Conventional treatment of early childhood caries in young children is an arduous task due to behavioral issue and lack of cooperative ability.

Hence, majority of patients are left untreated, causing premature loss of teeth paving them to cause psychological trauma, and hamper the timely eruption and normal occlusion in permanent teeth. From such point of view, the application of silver diamine fluoride, caries can be arrested, and above-described problems can be vanquished.⁶⁰

Once the carious process is halted, caries removal can be postponed to later stage when the child has the cooperative ability. SDF can be easily applied without any complicated tools and equipment, this allows treatment for children by allied staff under supervision in non-clinical environments, like outreach services in kindergartens, thus prevention at mass level.¹¹

Various studies mentioned below supports this implication. Hihara *et al.*⁶¹ in Japan, McDonald and Sheiham⁶² in London, Llodra *et al.* in Cuba,²⁹ Braga *et al.* in Brazil⁴¹ in Nepal found that SDF is significantly effective in arresting the cavitated as well as incipient carious lesions.

SDF is a colorless solution that can be applied on a tooth surface with a tiny applicator or brush. The concentration of SDF in commercial preparations varied from 3.8% to 38%. Preparation of 3.8% is specifically formulated for root canal treatment while 38% SDF is recommended for arrest of caries.⁶⁰

To prevent pit and fissure caries

Owing to the morphology and inaccessibility to clean, pits and fissures have higher susceptibility to dental caries. For the same reason, the topical fluoride application is revealed to be less effective for the prevention of the pit and fissure caries than that of the smooth surface. Due to its antibacterial and caries preventive property SDF has shown to be effective in the prevention of pits and fissures caries of the first molar teeth.⁶³

In an another study by Nishino and Massler, caries score of $\text{Ag}(\text{NH}_3)_2\text{F}$ treated teeth was significantly lower than the fissures treated with 8% SnF_2 or $\text{Ag}(\text{NO}_3)_3$.⁶⁴

Since SDF application also produces grayish black stain similar to incipient caries, recording of SDF application is mandatory.⁶⁵

To prevent secondary caries

For many decades, true adhesion has been the conundrum of dental restorative materials. No matter how much the restorative material adhere to the tooth structure completely; bacteria and food debris can always escape between the cavity walls and the restorative materials, making them more prone to recurrent caries. Thus, the resistance of cavity wall must be enhanced to prevent secondary caries.⁶⁶

SDF application on the cavity walls prior to amalgam restoration in primary teeth leads to complete eradication of recurrent caries for 26

months.⁴⁹ While SDF has shown no adverse effect on bonding of composite resin,⁶⁷ showed 50% decrease in dentin bond strength of GIC.⁶⁸

SDF reduced dentin bonding strength of resin based crown cement by $\sim 1/3$.⁶⁹ As a result, rinsing is advised for direct restorations, while removal of the SDF treated superficial dentin is recommended for cementing crowns.

To arrest root caries

Myriad of studies have mentioned increased incidence and prevalence of root caries with declining years.⁷⁰⁻⁷³ SDF, due its ability to arrest dental caries, annual application has been found to be quite effective in arresting root surface caries.^{42,74}

To desensitize sensitive teeth

Gottlieb,⁵⁰ suggested that the mechanism of arresting the dental caries is similar to that of desensitizing the hypersensitive dentin. Hence, SDF, owing to its ability to occlude dentinal tubules can be used to eliminate hypersensitivity.

A study by Kimura *et al.*⁷⁵ revealed that SDF was successful in eradicating sensitivity caused by erosion and abrasion. It was found that no more than four times application is needed for desensitizing effect.

To treat infected root canals

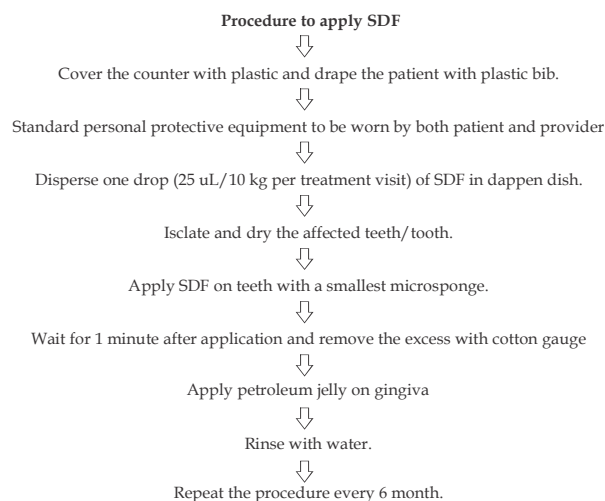
Tanaka conveyed that an aqueous solution of AgF has powerful disinfectant, protein-coagulation action; and occludes the dentinal tubules of root canal wall in terms of the electric resistance.⁷⁶ It has been mentioned that 3.8% SDF can be used as an eloquent root canal irrigant effectively exterminates the microbes present in the canal and circumpulpal dentin.^{77,78}

As an indirect pulp capping agent

As described by Yamaga *et al.*, if SDF is applied in the presence of softened dentin, it will arrest the subsequent progress of dental caries.¹²

SDF application in the presence of softened dentin has shown to arrest the progress of carious lesion. An *in vitro* study by Gupta *et al.*, manifested highest zone of bacterial inhibition around SDF.⁷⁹

Besides, an *in vivo* study quoted that SDF has remineralizing, re-hardening and antimicrobial virtue; thus, it can prove to be a prudent material for Indirect Pulp Capping.⁸⁰

Procedure⁸¹**Safety margin⁸¹**

- Average LD₅₀ by oral administration is 520 mg/kg, and by subcutaneous administration is 380 mg/kg. One drop (25 μ L), adequate to treat 5 teeth contains 9.5 mg SDF.
- Presume that the child patient is of 10 kg, the dose would be 0.95 mg/kg child.
- Hence, the relative safety margin of utilizing a full drop on a 10-kg child is: 380 mg/kg
- LD₅₀/0.95 mg/kg dose = 400-fold safety margin.

Unfavourable consequences⁸¹

- Among various interventional studies only a small, mildly painful white lesion in the mucosa which disappeared after 2 days without treatment, has been revealed.^{13,29,30,42}
- SDF causes darkening of the carious lesions.
- Patients may combat a short-term metallic or bitter taste.
- Little amount of SDF can result in “temporary tattoo” on skin with no harm. Stain on the skin fade away with the natural skin depilation, in 2–14 days.
- SDF stains the clinic surfaces and clothes. Spills should be cleaned up immediately with copious water, ethanol, or bleach.

Conclusion

Biannual application of 38% SDF has shown promising results over other minimal invasive techniques. Also, it is easy to use with minimal

side effects and wide array of safety margin. Nevertheless, it is obscure to predict the long-term outcome of such lesions. Further clinical trials are recommended to establish the same.

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